

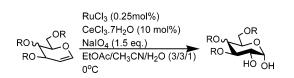
An Efficient Stereoselective Dihydroxylation of Glycals using a Bimetallic System, RuCl₃/CeCl₃/NaIO₄[†]

Pallavi Tiwari and Anup Kumar Misra*

Medicinal and Process Chemistry Division, Central Drug Research Institute, Chattar Manzil Palace, Lucknow-226 001, U.P., India

akmisra69@rediffmail.com

Received December 23, 2005



A catalytic dihydroxylation reaction on glycals has been developed using a bimetallic oxidizing system to furnish sugar 1,2-diols in a highly setreoselective manner.

Suitably protected 1,2-dihydroxy carbohydrate derivatives are useful synthetic intermediates in several organic transformations such as in the synthesis of polyhydroxylated chiral natural products,1 O-glycosides2 and C-glycosides.3 They have also used in the O-glycosylations involving intramolecular aglycon delivery.⁴ Conventionally, sugar-derived 1,2-diols are prepared using the reaction sequence of conversion of acetobromosugars to the corresponding sugar ortho esters followed by hydrolysis of ortho esters to the sugar 1,2-diols.⁵ Although this method has been used widely, use of excess s-collidine⁶ as solvent during the formation of ortho esters is a serious drawback of this protocol. Other reported methods for the preparation of sugar 1,2-diols include (a) osmium-catalyzed (OsO₄-NMO) dihydroxylation of glycals;^{2a,c} (b) conversion of glycals to 1,2glycal epoxides using dimethyldioxirane followed by hydrolytic opening of 1,2-glycal epoxides;7 and (c) reaction of glycals with

[†] C.D.R.I. communication no. 6921.

- (2) (a) Sanders, W. J.; Kiessling, L. L. Tetrahedron Lett. 1994, 35, 7335. (b) Shi, L.; Kim, Y.-J.; Gin, D. Y. J. Am.Chem. Soc. 2001, 123, 6939. (c) Charette, A. B.; Marcoux, J.-F.; Cote, B. Tetrahedron Lett. 1991, 32, 7215. (d) Trumtel, M.; Tavecchia, P.; Veyrières, A.; Sinay, P. Carbohydr. Res. 1989, 191, 29.
- (3) (a) Vidal, T.; Haudrechy, A.; Langlois, Y. Tetrahedron Lett. 1999, 40, 5677. (b) Hung, S.-C.; Wong, C.-H. Angew. Chem., Int. Ed. Engl. 1996, 35, 2671. (c) Carpintero, M.; Nieto, I.; Fernandez-Mayoralas, A. J. Org. Chem. 2001, 66, 1768.
- (4) (a) Barresi, F.; Hindsgual, O. J. Am. Chem. Soc. 1991, 114, 9376. (b) Stork, G.; Kim, G. J. Am. Chem. Soc. 1992, 114, 1087. (c) Bols, M. Chem. Commun. 1992, 913.

(5) (a) Lichtenthaler, F. W.; Schneider-Adams, T. J. Org. Chem. 1994, 59, 6728. (b) Broder, W.; Kunz, H. Carbohydr. Res. 1993, 249, 221. (c) Schmidt, R. R.; Effenberger, G. Carbohydr. Res. 1987, 171, 59. (d) Wu, E.; Wu, Q. Carbohydr. Res. 1993, 250, 327.

(6) Lemieux, R. U.; Morgan, A. R. Can. J. Chem. 1965, 43, 2199.

(7) (a) Danishefsky, S. J.; Halcomb, R. L. J. Am. Chem. Soc. 1989, 111, 6661. (b) Iserloh, U.; Dudkin, V.; Wang, Z.-G.; Danishefsky, S. J. *Tetrahedron Lett.* **2002**, *43*, 7027.

10.1021/jo0526385 CCC: \$33.50 © 2006 American Chemical Society Published on Web 02/25/2006

SCHEME 1

RO IN OR RO	RuCl ₃ (0.25mol%) CeCl ₃ .7H ₂ O (10 mol%) NalO ₄ (1.5 eq.) <u>EtOAc/CH₃CN/H₂O (3/3/1)</u> 0°C	ROM OR ROHOOH
		R = Acyl, alkyl

TABLE 1. Stereoselective cis-Dihydroxylation of Glycals and
Unsaturated Carbohydrate Derivatives using RuCl ₃ /CeCl ₃ ·7H ₂ O/
NaIO ₄ in EtOAc/CH ₃ CN/H ₂ O at 0 °C

Aco Aco Bzo Bzo Pivo Pivo Pivo OTBDPS Aco Aco Bno Bno Bno Bno Bno Bno Bno Bno Bno Bn	AcO OAc AcO OAc BZO OBZ BZO OBZ AcO OAc OPIV PIVO ACO OAc OTBDPS AcO OAc OBD BNO OBN BNO OBN BNO OBN BNO OBN BNO OBN BNO OBN BNO OBN	(min) 15 15 20 15 15 15 15	(%) 88 85 90 85 90 92	 2.6 α 2.75 1.9 1.0 1.6 	11 - - 8 11
Aco Bro Bro Aco Aco Bno Bno COBn Bno COBn Bno COBn Bno COBn Bno COBn COBn COBn COBn COBn COBn COBn COBn	AcO COAC BZO BZO ACO COAC PIVO PIVO ACO COAC OPIV PIVO ACO COAC COBN BNO ACO COAC COBN BNO ACO COAC ACO COAC COBN BNO ACO COAC COBN BNO ACO COAC COBN BNO ACO COAC COBN BNO ACO COAC COBN	15 20 15 15 15	85 90 85 90	α 2.75 1.9 1.0	-
BZO BZO PIVO PIVO OPIV OTBDPS ACO BNO ACO ACO BNO OBN BNO OBN BNO OBN BNO OBN BNO OBN BNO OBN BNO OBN OBN	BZO BZO ACO OPIV PIVO ACO OPIV PIVO ACO OTBDPS ACO OCBN BNO ACO OCBN BNO ACO OAC ACO OAC ACO OAC ACO OAC ACO OAC OBN BNO ACO OAC OBN OBN OBN OBN OCBN OCBN OCBN OCBN OC	20 15 15 15	90 85 90	2.75 1.9 1.0	- 8
Pivo O Pivo O OTBDPS AcO BnO AcO AcO BnO BnO OBn BnO OBn BnO OBn BnO OBn BnO OBn BnO OBn BnO OBn BnO O OBn BnO O O O O O O O O O O O O O	Pivo OPiv AcO OAc OTBDPS AcO OAc AcO OAc OBn BnO AcO OAc AcO OAc AcO OAc AcO OAc AcO OAc BnO OBn	15 15 15	85 90	1.9 1.0	- 8
AcO AcO BnO BnO AcO AcO BnO OBn BnO OBn BnO OBn	ACO OAC ACO OAC BNO ACO OAC ACO OAC ACO OAC ACO OAC ACO OAC BNO OBN	15 15	90	1.0	
Bno 29 Aco OAc Aco SBno OBn	Bno Bno Aco Aco Aco Aco Aco Aco Aco Aco Aco Ac	15			
Aco OBn OBn	AcO OAc AcO AcO OAc BNO OBn		92	1.6	11
10	BnO OBn	15			
			88	0.95	8
	AcO "OAc AcO_OZOAc AcO_OAc	15	85	9.8	-
AcO AcO OiPr	ACO OAC ACO OAC ACO OAC	20	90		-
Ph O O O OMe	Ph O OAc Aco	20	90		12
Aco Co Bno	AcO OAc AcO O O AcO O O BnO O	20	85		-
	Aco OAc Aco OAc	20	70 ^a		-
	OMe Aco COA Bno CoA Bno CoA Bno CoA	Ph O O O O O O O O O O O O O O O O O O O	Ph O Ac OMe AcO BnO BnO BnO BnO BnO BnO BnO COME AcO AcO AcO AcO AcO AcO AcO AcO	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$

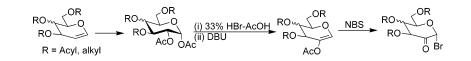
oxone in acetone.⁸ There are several drawbacks in the abovementioned methods, which include the use of very expensive and toxic reagents, difficulties in removing the Osmium salt from the products, use of very unstable epoxidation reagent, and formation of C-2 epimer. Use of oxone for the dihydroxylation of glycals is very convenient but always resulted in formation of a C-2 epimeric mixture as major and minor product in our hands. In view of the importance of 1,2-sugar diols in the synthesis of oligosaccharides and natural products, a strong impetus has been given to develop a mild, less toxic, economically convenient, and user-friendly reaction protocol for their preparation. Recently, we noted a few reports on the oxidative use of RuCl₃ in a combination of NaIO₄ and a Brønsted or Lewis

⁽¹⁾ Fürstner, A.; Konetzki, I. J. Org. Chem. 1998, 63, 3072.

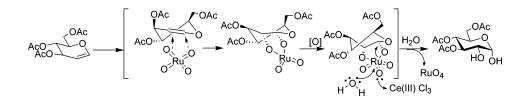
⁽⁸⁾ Rani, S.; Vankar, Y. D. Tetrahedron Lett. 2003, 44, 907.

JOC Note

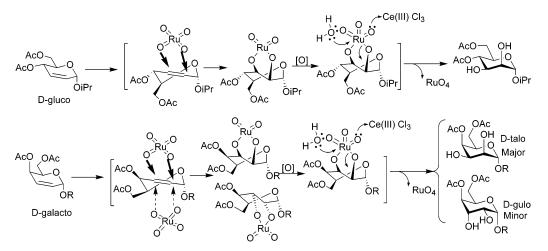
SCHEME 2



SCHEME 3



SCHEME 4



acid in a specific solvent mixture in the preparation of *cis*-1,2diols from simple alkenes in a highly stereoselective manner.⁹ We reasoned that use of RuCl₃/CeCl₃•7H₂O/NaIO₄, a bimetallic reagent system for the dihydroxylation of glycals, could produce only sugar 1,2-*cis*-diols as it does with the simple alkenes. In this endeavor, we report a highly stereoselective dihydroxylation of glycals toward the formation of 1,2-sugar diols (Scheme 1). This methodology has been further extended for the preparation of disaccharides of uncommon sugars (Table 1, entries 11 and 12).

In a first set of experiments, tri-O-acetyl-D-glucal was treated with RuCl₃/CeCl₃•7H₂O/NaIO₄ in EtOAc/CH₃CN/H₂O at 0 °C, varying the quantity of CeCl₃·7H₂O and NaIO₄. It was observed that treatment of tri-O-acetyl-D-glucal with a combination of RuCl₃/CeCl₃·7H₂O/NaIO₄ in a mixed solvent at 0 °C can furnish 3,4,6-tri-O-acetyl-D-glucose-1,2-diol in 15 min. A series of differentially protected glycals was transformed into 1,2-diols following the similar reaction condition. An anomeric mixture of sugar 1,2-diols was isolated, which were further acetylated for the structural confirmation of the products. From the spectral analysis, it was found that the stereochemistry at C-2 is equatorial in every case. In another set of experiments, a variety of 2,3-unsaturated mono- and disaccharides were also reacted with reagent combination toward the formation of 2,3unprotected mono- and disaccharides. Exclusive formation of cis-diol was observed depending on the stereochemistry of the substituents present in the substrates. As expected, no trace of the formation of other epimer was observed in TLC and spectral analysis. Under these reaction conditions, common protecting groups used in the protection of carbohydrates (e.g., benzylidene, isopropylidene, TBDPS, acetyl, benzyl, benzoyl, pivaloyl, etc.)

were quite stable. This stereoselective dihydroxylation has been applied to the preparation of disaccharides containing mannose, talose, and gulose using a reaction sequence consisting of Ferrier rearrangement followed by *cis*-dihydroxylation. Additionally, sugar 1,2-di-*O*-acetates thus obtained can be further converted to 2-acetoxy glycals by anomeric bromination using 33% HBr– AcOH and subsequent elimination of HBr on treatment with DBU. Synthesis of ulosyl bromides for their use in the glycosylation can also be prepared from the 2-acetoxy glycal or enol acetate by treatment with *N*-bromosuccinimide as reported by Lichtenthaler et al.¹⁰ (Scheme 2).

The plausible mechanism for the formation of exclusively single epimeric *cis*-diol can be explained by considering the *syn*-addition of the RuO_4 to the olefinic bond from the less sterically hindered site and hydrolysis of the Ru-complex activated by CeCl₃ (Schemes 3 and 4).

In summary, a mild, stereoselective protocol for the *cis*dihydroxylation of glycals and unsaturated sugar derivatives has been developed using an economically convenient, less toxic bimetallic catalyst system. The reaction is very fast and yields are excellent. The reaction protocol has been further extended toward the formation of 2-acetoxy glycal derivatives and unusual disaccharides.

^{(9) (}a) Plietker, B.; Niggemann, M. Org. Lett. **2003**, *5*, 3353. (b) Plietker, B.; Niggemann, M. Org. Biomol. Chem. **2004**, *2*, 1116. (c) Plietker, B.; Niggemann, M. J. Org. Chem. **2005**, *70*, 2402.

⁽¹⁰⁾ Lichtenthaler, F. W.; Schneider-Adams, T. J. Org. Chem. 1994, 59, 6728.

⁽¹¹⁾ Tai, A.-A.; Kulkarni, S. S.; Hung, S.-C. J. Org. Chem. 2003, 68, 8719.

⁽¹²⁾ Horton, D.; Luetzow, A. E. Carbohydr. Res. 1968, 7, 101.

Experimental Section

General Reaction Protocol for cis-Dihydroxylation of Glycal or 2,3-Unsaturated Glycoside. A mixture of NaIO₄ (300 mg, 1.4 mmol) and CeCl₃•7H₂O (38 mg, 0.1 mmol) in H₂O (2 mL) was stirred at room temperature for a few minutes. The reaction mixture was cooled to 0 °C, and to the cooled reaction mixture were added EtOAc (3 mL), CH₃CN (6 mL), and RuCl₃·H₂O (5.2 mg, 0.025 mmol) successively. After the mixture stirred for 2.0 min, a solution of substrate (1.0 mmol) in EtOAc (3 mL) was added, and the resulting heterogeneous mixture was stirred until the full consumption of the starting material (Table 1). After completion of the reaction (TLC), the reaction mixture was diluted with EtOAc (25 mL). The organic layer was washed with aqueous NaHCO3 and water, dried (Na₂SO₄), and concentrated under reduced pressure. The crude product was purified over SiO₂ using hexanes-EtOAc as eluant. Acetylation of the diol was carried out conventionally using acetic anhydride and pyridine, and the product was characterized by spectroscopic and analytical techniques. Spectral data of compounds **1b**, **5b**, **6b**, **7b**, and **10b** are reported in the cited references. Spectral data for all compounds are reported in Supporting Information.

Acknowledgment. Instrumentation facilities from SAIF, CDRI are gratefully acknowledged. P.T. thanks CSIR, New Delhi for providing a Senior Research fellowship. This project was funded by Department of Science and Technology (DST), New Delhi (SR/FTP/CSA-10/2002), India.

Supporting Information Available: General experimental methods, ¹H NMR and ¹³C NMR spectral data and spectra of all products, and 2D HSQC, HMBC, and COSY spectra of compounds **9b**, **11b**, and **12b**. This material is available free of charge via the Internet at http://pubs.acs.org.

JO0526385